



β -Metallation of Bridged Alkenyl Sulfones: Access to a Key Intermediate for Epibatidine Total Synthesis

Clifford D. Jones and Nigel S. Simpkins*

Department of Chemistry, University of Nottingham, University Park, Nottingham NG7 2RD, U.K.

and Gerard M. P. Giblin

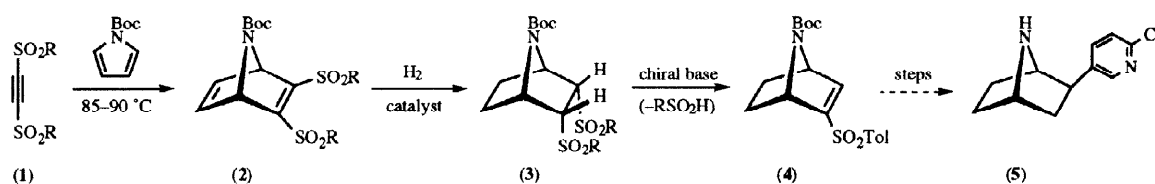
Medicinal Sciences, Glaxo Wellcome, Medicines Research Centre, Gunnels Wood Road, Stevenage, Hertfordshire, SG1 2NY England, U.K.

Received 21 October 1997; revised 20 November 1997; accepted 21 November 1997

Abstract: Efficient β -metallation of certain bridged alkenyl sulfones has been demonstrated. The reaction allows the synthesis of a *bis*-sulfone as a key intermediate for epibatidine synthesis.

© 1998 Elsevier Science Ltd. All rights reserved.

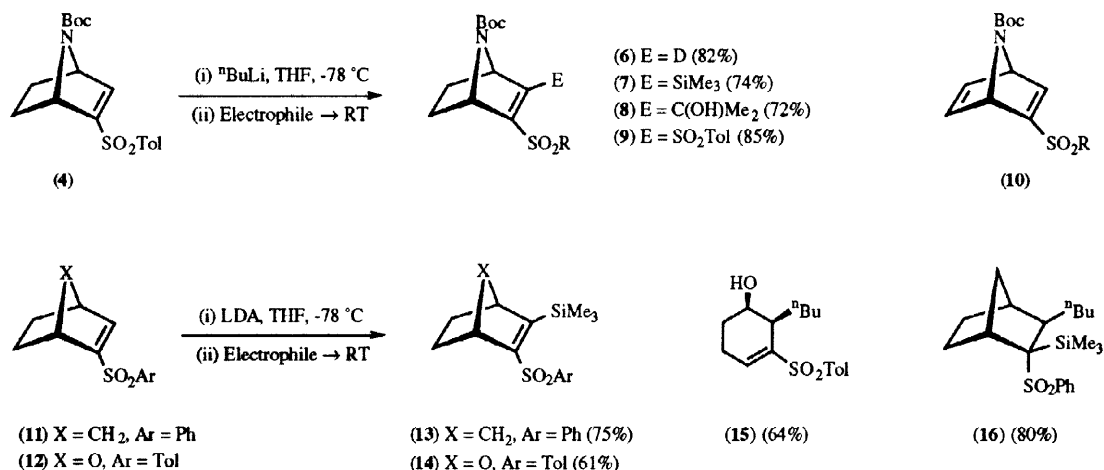
The alkaloid epibatidine **5**, possessed of both the unique 7-azabicyclo[2.2.1]heptane skeleton and intriguing analgesic properties, has been the subject of intense synthetic interest over the past few years.¹ We recently described a concise stereoselective synthesis of this compound in racemic form, by a route involving the Michael addition of a metallated pyridine to the key alkenyl sulfone **4**.² We conceived that this route would be amenable to a novel asymmetric variant whereby the alkenyl sulfone **4** would be accessed via chiral base mediated asymmetric elimination from a symmetrical *bis*-sulfone **3**.³



This in turn would be available via cycloaddition of alkyne **1** with *N*-Boc pyrrole to give **2**, followed by exhaustive hydrogenation.⁴ However, this plan soon ran into difficulties because the high reactivity of the *bis*-arylsulfonylalkyne **1** ($R = p\text{-Tol}$), precluded its use on a meaningful scale,⁵ whilst the alternative alkyne **1** ($R = t\text{Bu}$)⁶ gave a cycloadduct **2** with an alkenyl sulfone highly resistant to hydrogenation (presumably for steric reasons). This type of difficulty has been noted previously; the use of chlorinated alkenyl *bis*-sulfones, as described by De Lucchi and co-workers, providing one solution.⁷

Herein we demonstrate an alternative solution to the problem, which relies on the β -metallation of alkenyl sulfones, such as **4** (themselves very readily available by cycloadditions of ethynyl tolyl sulfone) to allow the introduction of several different substituents, including the sulfonyl group required for conversion into **3**. Although this mode of sulfone metallation has previously been noted as an unwanted pathway competing with organometallic addition to certain alkenyl sulfones,⁸ this reaction has not been demonstrated to be synthetically useful, and indeed, we expected that predominant metallation of the aromatic nucleus could be a problem.⁹

Therefore, we were delighted to find that treatment of alkenyl sulfone **4** with $^n\text{BuLi}$ at $-78\text{ }^\circ\text{C}$ resulted in clean metallation at the β -position, as evidenced by the recovery of fully deuterated material **6** on quenching the reaction mixture with D_2O . As shown, use of Me_3SiCl , acetone or $p\text{-TolSO}_2\text{F}$ as the electrophile resulted in efficient formation of **7**, **8** and **9** respectively; preparation of **9** by this route solving our problem in accessing *bis*-sulfone **3** through cycloaddition of **1** (see accompanying paper).



Attempts to carry out analogous reaction sequences using the unsaturated analogue **10** were not successful, whilst reactions involving attempted metallation of the carbocyclic **11**, and oxacyclic **12**, analogues (using Me_3SiCl as the electrophile), led instead to the addition products **15** and **16** respectively. However, by changing the base from $^n\text{BuLi}$ to LDA both of these systems could be metallated in the desired fashion, to give acceptable yields of the silylated products **13** and **14**. Further exploration of this mode of metallation, along with applications to target synthesis are underway.

Acknowledgements

We are grateful to the Engineering and Physical Sciences Research Council (EPSRC) and Glaxo Wellcome for support of C. D. J. under the CASE scheme.

References and Footnotes

- For the most recent work, see Kosugi, H.; Abe, M.; Hatsuda, R.; Uda, H.; Kato, M. *J. Chem. Soc., Chem. Commun.* **1997**, 1857.
- Giblin, G. M. P.; Jones, C. D.; Simpkins, N. S. *Synlett* **1997**, 589.
- See accompanying paper for preliminary results of the enantioselective elimination reaction.
- The apparent alternative, involving the use of an alkenyl *bis*-sulfone is not viable because these compounds do not undergo cycloaddition with pyrrole partners.
- Pasquato, L.; De Lucchi, O.; Krotz, L. *Tetrahedron Lett.* **1991**, 32, 2177.
- Riera, A.; Marti, M.; Moyano, A.; Pericas, M. A.; Santamaria, J. *Tetrahedron Lett.* **1990**, 31, 2173.
- De Lucchi, O.; Cossu, S. *J. Chem. Soc., Chem. Commun.* **1992**, 1089.
- (a) Barton, D. L.; Conrad, P. C.; Fuchs, P. L. *Tetrahedron Lett.* **1980**, 21, 1811. (b) Cassidy, J. F.; Williams, J. M. *Tetrahedron Lett.* **1986**, 27, 4355. (c) Fuchs, P. L.; Braish, T. F. *Chem. Rev.* **1986**, 903.
- For a review of sulfone metallations, see Simpkins, N. S. *Sulfones in Organic Synthesis*, Pergamon Press, 1993.